

COURSE OUTLINE

(1) GENERAL

SCHOOL	SCHOOL OF SCIENCES		
DEPARTMENT	DEPARTMENT OF CHEMISTRY		
LEVEL OF STUDIES	ISCED level 6 – Bachelor’s or equivalent level		
COURSE CODE	YN305	SEMESTER	3rd Semester
COURSE TITLE	English II		
TEACHING ACTIVITIES <i>If the ECTS Credits are distributed in distinct parts of the course e.g. lectures, labs etc. If the ECTS Credits are awarded to the whole course, then please indicate the teaching hours per week and the corresponding ECTS Credits.</i>	TEACHING HOURS PER WEEK	ECTS CREDITS	
	3	2	
<i>Please, add lines if necessary. Teaching methods and organization of the course are described in section 4.</i>			
COURSE TYPE <i>Background, General Knowledge, Scientific Area, Skill Development</i>	Skill Development		
PREREQUISITES:	NO		
TEACHING & EXAMINATION LANGUAGE:	ENGLISH		
COURSE OFFERED TO ERASMUS STUDENTS:	NO		
COURSE URL:	https://eclass2.emt.duth.gr/courses/CHEM-N3105/		

(2) LEARNING OUTCOMES

Learning Outcomes <i>Please describe the learning outcomes of the course: Knowledge, skills and abilities acquired after the successful completion of the course.</i>	
<p>The course aims to help students use English correctly and accurately for academic and professional purposes. Upon successful completion of the course, students are expected to have improved their language skills in English and acquired basic techniques to be able to:</p> <ul style="list-style-type: none"> • Study and use English scientific literature for the preparation of assignments, seminars, etc. • Understand academic texts and the corresponding terminology • Write summaries of English texts in Greek • Write a scientific text in English • Attend lectures in English • Give presentations in English 	
General Skills <i>Name the desirable general skills upon successful completion of the module</i>	
<i>Search, analysis and synthesis of data and information, ICT Use</i>	<i>Project design and management</i>
<i>Adaptation to new situations</i>	<i>Equity and Inclusion</i>
<i>Decision making</i>	<i>Respect for the natural environment</i>
<i>Autonomous work</i>	<i>Sustainability</i>
<i>Teamwork</i>	<i>Demonstration of social, professional and moral responsibility and sensitivity to gender issues</i>
<i>Working in an international environment</i>	<i>Critical thinking</i>
<i>Working in an interdisciplinary environment</i>	<i>Promoting free, creative and inductive reasoning</i>
<i>Production of new research ideas</i>	
Search, analysis and synthesis of data and information, Autonomous work, Teamwork, Working in an international environment	

(3) COURSE CONTENT

Week 1: Separation techniques, distillation, evaporation.
Week 2: Crystallization, thin layer or paper chromatography, separating funnel, filtration.
Week 3: Acids and Bases.
Week 4: Electricity, electrochemistry.
Week 5: Electroplating, electrolysis, redox reactions, galvanic cells.
Week 6: Water treatment methods.
Week 7: Distillation, pre and post chlorination, reverse osmosis.
Week 8: Polymers, synthetic, natural, thermosetting, thermoplastic, linear, branched, cross-linked, fibers.
Week 9: Plastics.
Week 10: Elastomers.
Week 11: Terminology in Organic chemistry.
Week 12: Terminology in Inorganic chemistry.
Week 13: Terminology in Analytical chemistry.

(4) LEARNING & TEACHING METHODS - EVALUATION

TEACHING METHOD <i>Face to face, Distance learning, etc.</i>	Face to face												
USE OF INFORMATION & COMMUNICATIONS TECHNOLOGY (ICT) <i>Use of ICT in Teaching, in Laboratory Education, in Communication with students</i>	Use of ICT in Teaching Use of ICT in Communication with students												
TEACHING ORGANIZATION <i>The ways and methods of teaching are described in detail.</i> <i>Lectures, Seminars, Laboratory Exercise, Field Exercise, Bibliographic research & analysis, Tutoring, Internship (Placement), Clinical Exercise, Art Workshop, Interactive learning, Study visits, Study / creation, project, creation, project. Etc.</i> <i>The supervised and unsupervised workload per activity is indicated here, so that total workload per semester complies to ECTS standards.</i>	<table border="1"><thead><tr><th>Activity</th><th>Workload/semester</th></tr></thead><tbody><tr><td>Lectures</td><td>39</td></tr><tr><td>Homework</td><td>13</td></tr><tr><td>Study</td><td>13</td></tr><tr><td>Exams</td><td>3</td></tr><tr><td>Total</td><td>68</td></tr></tbody></table>	Activity	Workload/semester	Lectures	39	Homework	13	Study	13	Exams	3	Total	68
	Activity	Workload/semester											
	Lectures	39											
	Homework	13											
	Study	13											
	Exams	3											
Total	68												
STUDENT EVALUATION <i>Description of the evaluation process</i> <i>Assessment Language, Assessment Methods, Formative or Concluding, Multiple Choice Test, Short Answer Questions, Essay Development Questions, Problem Solving, Written Assignment, Essay / Report, Oral Exam, Presentation in audience, Laboratory Report, Clinical examination of a patient, Artistic interpretation, Other/Others</i> <i>Please indicate all relevant information about the course assessment and how students are informed</i>	Students are assessed by means of a final written exam in English.												

(5) SUGGESTED BIBLIOGRAPHY

- 1. English for Chemistry EAP, Κατσαμποξάκη-Hodgetts Κάλλια, Εκδόσεις Δίσιγμα, 2017
- 2. The Chemistry Between Us, Βασιλική Ριζομυλιώτη, Ζωή Μαλιβίτση, Εκδόσεις Αθανάσιου Αλτιντζή, 2021