

COURSE OUTLINE

(1) GENERAL

SCHOOL	SCHOOL OF SCIENCES		
DEPARTMENT	DEPARTMENT OF CHEMISTRY		
LEVEL OF STUDIES	ISCED level 6 – Bachelor's or equivalent level		
COURSE CODE	EN23	SEMESTER	7th Semester
COURSE TITLE	Inorganic Materials Chemistry		
TEACHING ACTIVITIES	<p>If the ECTS Credits are distributed in distinct parts of the course e.g. lectures, labs etc. If the ECTS Credits are awarded to the whole course, then please indicate the teaching hours per week and the corresponding ECTS Credits.</p>		TEACHING HOURS PER WEEK
			3
Please, add lines if necessary. Teaching methods and organization of the course are described in section 4.			
COURSE TYPE	Skill Development		
Background, General Knowledge, Scientific Area, Skill Development			
PREREQUISITES:	NO		
TEACHING & EXAMINATION LANGUAGE:	GREEK		
COURSE OFFERED TO ERASMUS STUDENTS:			
COURSE URL:	https://eclass2.emt.duth.gr/courses/CHEM_H110/		

(2) LEARNING OUTCOMES

Learning Outcomes																			
Please describe the learning outcomes of the course: Knowledge, skills and abilities acquired after the successful completion of the course.																			
Upon completion of the course, students will be able to:																			
<ul style="list-style-type: none"> - understand the concept of crystal structure in solid materials. - understand the basic principles of the chemical structure of crystalline solids. - understand the importance of imperfections in solids. - understand the concept of dislocations and strengthening mechanisms in metals. - understand phase diagrams of materials. - understand the chemical properties and applications of metals and ceramics. - understand the chemistry of composite materials. 																			
General Skills																			
Name the desirable general skills upon successful completion of the module																			
<table style="width: 100%; border-collapse: collapse;"> <tr> <td style="width: 50%;">Search, analysis and synthesis of data and information,</td><td style="width: 50%;">Project design and management</td></tr> <tr> <td>ICT Use</td><td>Equity and Inclusion</td></tr> <tr> <td>Adaptation to new situations</td><td>Respect for the natural environment</td></tr> <tr> <td>Decision making</td><td>Sustainability</td></tr> <tr> <td>Autonomous work</td><td>Demonstration of social, professional and moral responsibility and sensitivity to gender issues</td></tr> <tr> <td>Teamwork</td><td>Critical thinking</td></tr> <tr> <td>Working in an international environment</td><td>Promoting free, creative and inductive reasoning</td></tr> <tr> <td>Working in an interdisciplinary environment</td><td></td></tr> <tr> <td>Production of new research ideas</td><td></td></tr> </table>		Search, analysis and synthesis of data and information,	Project design and management	ICT Use	Equity and Inclusion	Adaptation to new situations	Respect for the natural environment	Decision making	Sustainability	Autonomous work	Demonstration of social, professional and moral responsibility and sensitivity to gender issues	Teamwork	Critical thinking	Working in an international environment	Promoting free, creative and inductive reasoning	Working in an interdisciplinary environment		Production of new research ideas	
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Search, analysis and synthesis of data and information, Adaptation to new situations, Decision making, Autonomous work, Teamwork, Working in an international environment, Working in an interdisciplinary environment, Production of new research ideas, Project design and management, Critical thinking, Promoting free, creative and inductive reasoning																			

(3) COURSE CONTENT

Week 1: Introduction to the course and material properties.
 Week 2: Basic crystallography.
 Week 3: Chemical structure of crystalline solids - metals.
 Week 4: Chemical structure of crystalline solids - ceramics.
 Week 5: Solid imperfections - metals.
 Week 6: Solid imperfections - ceramics.
 Week 7: Dislocations.
 Week 8: Metals strengthening mechanisms.
 Week 9: Phase diagrams.
 Week 10: Phase transitions.
 Week 11: Chemical properties and applications of metals.
 Week 12: Chemical properties and applications of ceramics.
 Week 13: Chemistry of composite materials.

(4) LEARNING & TEACHING METHODS - EVALUATION

TEACHING METHOD <i>Face to face, Distance learning, etc.</i>	Face to face	
USE OF INFORMATION & COMMUNICATIONS TECHNOLOGY (ICT) <i>Use of ICT in Teaching, in Laboratory Education, in Communication with students</i>	Use of ICT in Teaching Use of ICT in Communication with students	
TEACHING ORGANIZATION <i>The ways and methods of teaching are described in detail. Lectures, Seminars, Laboratory Exercise, Field Exercise, Bibliographic research & analysis, Tutoring, Internship (Placement), Clinical Exercise, Art Workshop, Interactive learning, Study visits, Study / creation, project, creation, project. Etc. The supervised and unsupervised workload per activity is indicated here, so that total workload per semester complies to ECTS standards.</i>	Activity	Workload/semester
	Lectures	39
	Study	33
	Exams	3
	Total	75
STUDENT EVALUATION <i>Description of the evaluation process Assessment Language, Assessment Methods, Formative or Concluding, Multiple Choice Test, Short Answer Questions, Essay Development Questions, Problem Solving, Written Assignment, Essay / Report, Oral Exam, Presentation in audience, Laboratory Report, Clinical examination of a patient, Artistic interpretation, Other/Others Please indicate all relevant information about the course assessment and how students are informed</i>	Written final exam that includes problem solving from different sections of the course (no notes allowed).	

(5) SUGGESTED BIBLIOGRAPHY

- Επιστήμη και Τεχνολογία των Υλικών, 10η Έκδοση, Callister William D., Rethwisch David G.
- Υλικά, δομή, ιδιότητες και Τεχνολογικές Εφαρμογές, 7η Έκδοση, Askeland Donald, Wendelin Wright
- Επιστήμη και τεχνολογία υλικών, Βατάλης Αργύρης Σ.
- Χημεία και Τεχνολογία Υλικών, Ignatowitz E., Fischer U.